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elements, and molecules  
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electron calculation  
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~~Structure: Protons,~~

~~Neutrons, Electrons,~~

~~Mass, Atomic Number~~



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Structure and Key

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Atomic Structure  
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1 a) protons b) neutrons  
c) electrons a) Positive b)

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Neutral c) negative 2  
atomic number (or  
identity); charge 3  
protons; electrons (in a  
neutral charge atom  
only!); same 4 average  
atomic weight; mass 5  
mass number; nucleus 6  
neutrons; protons (or  
atomic number); mass  
number 7 Lithium = Li  
= 3 Iron = Fe=26  
Oxygen = O = 8  
Krypton = Kr = 36 8 (A

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neutral atom by  
definition has NO  
charge!

~~Basic Atomic Structure  
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Worksheet H and the 1.

2. 4. 5. 6. 8. 9. The 3  
particles of the atom  
are: protons ( $e^+$ ) (e) Their  
respective charges are:

The number of protons  
in one atom of an

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Atomic structure worksheet answers key

element determines the atom's number Of electrons determines the of the elemen. in one atom of an element.

~~090412 Atomic Structure Worksheet 1~~  
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1 a) protons b) neutrons  
c) electrons a) Positive b) Neutral c) negative 2  
atomic number (or

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identity); charge 3  
protons; electrons (in a  
neutral charge atom  
only!); same 4 average  
atomic weight; mass 5  
mass number; nucleus 6  
neutrons; protons (or  
atomic number); mass  
number 7 Lithium = Li  
= 3 Bromine = Br = 35  
Iron = Fe = 26 Copper =  
Cu = 29 Oxygen = O =  
8 Mercury = Hg = 80  
Krypton = Kr = 36

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Helium = He = 2 8 (A neutral atom by definition has NO charge!

Answers Key

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Write two algebraic equations relating the number of protons and neutrons in an atom to

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that atom's atomic mass and atomic number: Where,  $P =$  Number of protons.  $N =$  Number of neutrons.  $A =$  Atomic mass.  $Z =$  Atomic number

~~Atomic Structure~~  
~~Worksheet - Basic~~  
~~Electricity~~

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Worksheet and the 1. 2.  
3. 4. 5. 6. 8. 9. The 3

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particles of the atom are: Their respective charges are: The number of protons in one atom of an element determines the atom's number of electrons determines the of the element. The atomic number tells you the number of in one atom of an element. It also tells you the number of

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10. Name the element which has the following numbers of particles: a. b. c. d. 26 electrons, 29 neutrons, 26 protons 53 protons, 74 neutrons 2 electrons (neutral atoms)

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Atomic Basics Answer

Key Part A: Atomic

Structure 1. Draw five

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protons in the nucleus of the atom. Label them with their charge. 2.

Draw six neutrons in the nucleus of the atom. 3.

Draw two electrons in the first energy level and label them with their charge. 4. Draw three electrons in the second energy level and label them with their charge.

5.

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~~Chemistry of Matter~~

Basic atomic structure  
worksheet h and the 1.

Chlorine 35 42 atomic  
number equals the

number of or atomic  
mass equals the number  
of identify the each of  
the parts of the box.

Their respective charges  
are.

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3. The atomic number tells you the number of \_\_\_\_\_ in one atom of an element. It also tells you the number of \_\_\_\_\_ in a neutral atom of that element. The atomic number gives the “ identity ” of an element as well as its location on the periodic table. No two different elements will

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of Atomic Structure,

Atomic Number, Mass

Number, Symbols,

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Challenge, and  
questions like Label the  
parts of an atom on the  
diagram below. Atomic  
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Diagrams, ....

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$9.1095 \times 10^{-28}$  K g.

Unit electrical charge is

$1.6022 \times 10^{-19}$

coulomb (C). The

nucleus at the center of

the atom contains one

or more positively

charged protons. All

atoms of a given

element have the same

number of protons,

which defines the

element's atomic

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number, given the  
symbol  $Z$ .

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worksheet answers. It is  
composed of protons  
neutrons and electrons.  
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protons, neutrons and

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electrons. The majority of an atom is empty space. The nucleus contains the majority of an atom's mass and is incredibly small. If the atom was a size of a sports arena the nucleus would be as small as a pea! The hydrogen atom is the most abundant type of atom in the universe.

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